

Chapter 3 The Mole And Stoichiometry Part 1 The Mole

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how do you define stoichiometry stoichiometry is the branch of chemistry that deals with the relationship between the relative quantities of substances taking part in a chemical reaction

this is called stoichiometry which deals with figuring out the amount of products if you are given a certain amount of reactants or figuring out how much reactants you need to get a certain amount of

watch a step by step example to understand the process involved in mass to mass stoichiometry learn to convert between the masses of reactants and products using balanced equations and molar masses

now that we have the balanced equation let's get to problem solving to review we want to find the mass of naoh that is needed to completely react 3 10 grams of h a 2 so a 4 we can tackle this

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